

	Chemical name (Synonym)	Chemical formula (CAS number)	Concentration	20 °C 68 °F	Other
	Carbonic acid	H <sub>2</sub> CO <sub>3</sub> (463-79-6)	-	Ex	-
	Fluorosilicic acid	H <sub>2</sub> SiF <sub>6</sub> (16961-83-4)	-	М	-
	Hydrobromic acid	HBr (10035-10-6)	5%	Ex	-
cids	Hydrochloric acid	HCl (7647-01-0)	36% 20% 10% 5%	P M G Ex	Pass 72 hrs in 36% - -
Inorganic Acids	Nitric acid	HNO <sub>3</sub> (7697-37-2)	20% 10% 5%	M G Ex	- - -
드	Nitrous acid	HNO <sub>2</sub> (7782-77-6)	10%	Ex	-
-	Oleum Phosphoric acid (orthophosphoric acid)	H <sub>3</sub> PO <sub>4</sub> (7664-38-2)	- 85% 10%	P P G	- Pass 72 hrs in 85%
-	Sulfuric acid	H <sub>2</sub> SO <sub>4</sub> (7664-93-9)	98% 20% 10% 5%	P M G Ex	Pass 72 hrs in 50% - - -
	Acetic acid (ethanoic acid)	CH <sub>3</sub> COOH (64-19-7)	20%	P M	Pass 72 hrs in 10%
	Chlorosulfonic acid (sulfurochloridic acid)	HSO <sub>3</sub> Cl (7790-94-5)	-	М	-
cids	Citric acid	C <sub>6</sub> H <sub>8</sub> O <sub>7</sub> (77-92-9)	-	G	-
Organic Acids	Cresylic acid (cresol)	C <sub>7</sub> H <sub>8</sub> O (1319-77-3)	-	Р	-
Org	Formic acid (methanoic acid)	HCOOH (64-18-6)	20% 5%	P M	- -
	Lactic acid (2-hydroxypropanoic acid)	CH <sub>3</sub> CH(OH)(COOH) (50-21-5/79-33-4/10326-41-7)	10%	М	-
	Phenol	$C_6H_5OH$ (108-95-2)	80%	Р	-
	n-Butanol (butyl alcohol)	C <sub>4</sub> H <sub>9</sub> OH (71-36-3)	-	Ex	-
	Ethanol (ethyl alcohol)	CH <sub>3</sub> CH <sub>2</sub> OH (64-17-5)	-	М	-
hols	Ethylene glycol (ethan-1,2-diol, monoethylene glycol, MEG)	(CH <sub>2</sub> OH) <sub>2</sub> (107-21-1)	-	Ex	-
Alcohols	Glycerol (glycerine, propane-1,2,3-triol)	HOCH <sub>2</sub> CH(OH)CH <sub>2</sub> OH (56-81-5)	-	Ex	-
	Higher alcohols	$C_nH_{(2n+1)}OH$ where $n > 2$	-	Ex	-
	Methanol (methyl alcohol)	CH <sub>3</sub> OH (67-56-1)	-	Р	Pass 72 hrs
	2-Methoxyethanol	$C_3H_8O_2$ (109-86-4)	-	Ex	-

Excellent	Ex	Suitable for all reasonable applications including immersion.	
Good	G	Suitable for applications involving immersion for short periods, splashing and contact with fumes.	
Moderate	М	Suitable for use in environments contaminated by the chemical or in situations where accidental splashing can be removed either by cleaning or in the case of volatile solvents, by evaporation.	
Poor	Р	Not suitable for any applications involving contact with the chemical itself or fumes evolved from it.	
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	Chemical name (Synonym)	Chemical formula (CAS number)	Concentration	20 °C 68 °F	Other
s	Propan-1-ol (Propyl alcohol)	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> OH (71-23-8)	-	Ex	-
Alcohols	Propylene glycol (1,2-Propanediol)	CH <sub>3</sub> CH(OH)CH <sub>2</sub> OH (57-55-6)	-	Ex	-
₹ 9	Secondary alcohols	R₁R₂CHOH	=	Ex	-
	Tertiary alcohols	R₁R₂R₃COH	-	Ex	-
	Ammonia	NH <sub>3</sub> (7664-41-7)	20% 10%	Ex Ex	-
	Barium hydroxide	Ba(OH) <sub>2</sub> (17194-00-2)	-	Ex	-
	Calcium hydroxide (lime water)	Ca(OH) <sub>2</sub> (1305-62-0)	-	Ex	-
Alkalis	Magnesium hydroxide (milk of magnesia)	Mg(OH) <sub>2</sub> (1309-42-8)	-	Ex	-
₹	Potassium hydroxide		40%	Ex	-
	(caustic potash)	KOH (1310-58-3)	20%	Ex	-
	(caustic potasti)	(1310-36-3)	10%	Ex	-
			50%	Ex	
	Sodium hydroxide	NaOH	40%	Ex	-
	(caustic soda)	(1310-73-2)	20%	Ex	-
			10%	Ex	-
	Aniline (Phenylamine)	C <sub>6</sub> H <sub>5</sub> NH <sub>2</sub> (62-53-3)	-	M	-
S	Diethanolamine	HN(CH <sub>2</sub> CH <sub>2</sub> OH) <sub>2</sub> (111-42-2)	-	Ex	-
Amines & Amides	Diethylamine	CH <sub>3</sub> CH <sub>2</sub> NHCH <sub>2</sub> CH <sub>3</sub> (109-89-7)	-	Р	-
s & /	Dimethylformamide	(CH <sub>3</sub> ) <sub>2</sub> NC(O)H (68-12-2)	-	Р	-
mine	Methylamine (25% in water)	CH <sub>3</sub> NH <sub>2</sub> (74-89-5)	-	G	-
,	Pyridine	C <sub>5</sub> H <sub>5</sub> N (110-86-1)	-	Р	-
	Triethanolamine (TEA) (2,2',2"-nitrilotriethanol)	N(CH <sub>2</sub> CH <sub>2</sub> OH) <sub>3</sub> (102-71-6)	-	Ex	-
Ņ	Beer		-	Ex	-
Foodstuffs	Cider		-	Ex	-
dst	Citrus juices		-	Ex	-
F00	Fermentation liquor		-	Ex	-
∞ಶ	Glucose		-	Ex	-
ges	Milk		-	G	-
era	Sugar solution		-	Ex	-
Beverages	Vinegar		=	М	=
Ш	Whisky and Wine		-	Ex	-

Excellent	Ex	Suitable for all reasonable applications including immersion.	
Good	G	Suitable for applications involving immersion for short periods, splashing and contact with fumes.	
Moderate	Suitable for use in environments contaminated by the chemical or in situations where accidental splashing can be removed either by cleaning or in the case of volatile solvents, by evaporation.		
Poor	Р	Not suitable for any applications involving contact with the chemical itself or fumes evolved from it.	
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	Chemical name (Synonym)	Chemical formula (CAS number)	Concentration	20 °C 68 °F	Other
	Amyl acetate	CH <sub>3</sub> COO(CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub> (628-63-7)	-	G	-
	Butyl acetate	C <sub>6</sub> H <sub>12</sub> O <sub>2</sub> (123-86-4)	-	G	-
	Dibutyl adipate	[CH <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub> ] <sub>2</sub> (105-99-7)	-	Ex	-
	Dibutyl phthalate	C <sub>16</sub> H <sub>22</sub> O <sub>4</sub> (84-74-2)	-	Ex	-
s	Dibutyl sebacate	C <sub>18</sub> H <sub>34</sub> O <sub>4</sub> (109-43-3)	-	Ex	-
Esters & Ethers	Diethyl ether	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O (60-29-7)	-	Ex	
rs &	Dioctyl adipate	C <sub>22</sub> H <sub>42</sub> O <sub>4</sub> (123-79-5)	-	Ex	-
Este	Dioctyl phthalate	C <sub>6</sub> H <sub>4</sub> (C <sub>8</sub> H <sub>17</sub> COO) <sub>2</sub> (117-81-7)	-	Ex	-
	Dioctyl sebacate	(CH <sub>2</sub> ) <sub>8</sub> (COOC <sub>8</sub> H <sub>17</sub> ) <sub>2</sub>	-	Ex	-
	Diphenyl isodecyl phosphate	C <sub>22</sub> H <sub>31</sub> O <sub>4</sub> P (29761-21-5)	-	Ex	-
	Ethyl acetate	CH <sub>3</sub> COOCH <sub>2</sub> CH <sub>3</sub> (141-78-6)	-	G	-
	Isopropyl ether	C <sub>6</sub> H <sub>14</sub> O (108-20-3)	-	Ex	-
	Methyl acetate	CH <sub>3</sub> COOCH <sub>3</sub> (79-20-9)	-	G	-
	Carbon dioxide (dry)	CO <sub>2</sub> (124-38-9)	-	Ex	-
	Carbon monoxide	CO (630-08-0)	-	Ex	-
	Chlorine (dry)	Cl <sub>2</sub> (7782-50-5)	-	Ex	-
	Chlorine (wet)	Cl <sub>2</sub> (7782-50-5)	-	M	-
	Hydrogen	H <sub>2</sub> (1333-74-0)	-	Ex	-
Gases	Natural Gas (Methane)	CH <sub>4</sub>	-	Ex	-
g	Nitrogen	N <sub>2</sub> (7727-37-9)	-	Ex	-
	Nitrous oxide (dinitrogen monoxide)	N <sub>2</sub> O (10024-97-2)	-	Ex	-
	Ozone (dry)	O <sub>3</sub> (10028-15-6)	-	Ex	-
	Ozone (aqueous solution)		-	М	-
	Sulphur dioxide	SO <sub>2</sub> (7446-09-5)	-	Ex	-
	Sulphur trioxide (sulphuric anhydride)	SO <sub>3</sub> (7446-11-9)	-	Ex	-

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	Chemical name (Synonym)	Chemical formula (CAS number)	Concentration	20 °C 68 °F	Other
	Carbon tetrachloride	CCl <sub>4</sub> (56-23-5)	-	G	-
	Chlorobenzene	C <sub>6</sub> H <sub>5</sub> Cl (108-90-7)	-	M	-
Halocarbons	Chloroform	CHCl <sub>3</sub> (67-66-3)	-	G	-
arb	Dry cleaning fluids		-	G	-
loc	Methylene chloride	CH <sub>2</sub> Cl <sub>2</sub>	-	Р	-
На	(dichloromethane)	(75-09-2)			
	Perchloroethylene	$Cl_2C=CCl_2$	-	G	-
	(tetrachloroethylene)	(127-18-4)			
	1,1,1, - Trichloroethane (methyl chloroform)	CH <sub>3</sub> CCl <sub>3</sub> (71-55-6)	-	G	=
	Aviation fuel				
	(AVCAT, AVGAS, AVTAG, AVTUR)	N/A	-	Ex	-
	Benzene	C <sub>6</sub> H <sub>6</sub>			
	(benzol)	(71-43-2)	-	M	-
	Cyclohexane	C <sub>6</sub> H <sub>12</sub> (110-82-7)	-	G	-
	Gasoline (Petrol)		-	G	-
	Heptane	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> (142-82-7)	-	Ex	-
SI	Hexane	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> (110-54-3)	-	Ex	-
Hydrocarbons	lso-octane (2,2,4-trimethylpentane)	(CH <sub>3</sub> ) <sub>3</sub> CCH <sub>2</sub> CH(CH <sub>3</sub> ) <sub>2</sub> (540-84-1)	-	Ex	-
lydro	Kerosene	N/A (8008-20-6)	-	Ex	-
-	Paraffin	N/A (8002-74-2)	-	Ex	-
	Pentane	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> (109-66-0)	-	Ex	-
	Styrene	$C_6H_5CH=CH_2$ (100-42-5)	-	G	-
	Toluene (methylbenzene, phenylmethane, toluol)	C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub> (108-88-3)	-	Р	-
	White Spirit (Stoddard solvent, Mineral spirits)	(8052-41-3)	-	Ex	-
	Xylene (dimethyl benzene, xylol)	C <sub>6</sub> H <sub>4</sub> (CH <sub>3</sub> ) <sub>2</sub> (95-47-6/108-38-3/106-42-3/1330-20-7)	-	M	-
nes	Acetone	(CH <sub>3</sub> ) <sub>2</sub> CO (67-64-1)	-	M	-
Ketones	Methyl ethyl ketone (MEK, butanone)	CH <sub>3</sub> C(O)CH <sub>2</sub> CH <sub>3</sub> (78-93-3)	-	M	-
S	Bleach	_	-	М	-
) on	Brake fluid		=	Ex	=
ane	Drilling mud		-	Ex	-
Cell	Emulsion paint		-	Ex	-
Miscellaneous	Fertilizer solutions		-	Ex	-
	Grease		-	Ex	-

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	Ink (water based)		-	Ex	-
sno	Mercury	Hg	-	Ex	-
Miscellaneous continued	Mine waters (acid)		-	Ex	-
ella ottio	Oil/water mixtures		-	Ex	-
isc	Water, distilled		-	Ex	-
≥	Water, fresh		-	Ex	-
	Water, sea		-	Ex	-
<u>.</u>	Bunker oils (fuel oils)  Crude oil		-	Ex Ex	-
le l	Crude oil  Cutting oils, water emulsions		-	Ex	-
- Mineral	Diesel oil		-	Ex	-
oils -	Lubricating oil		-	Ex	-
	Transformer oil		-	Ex	-
	Castor oil		=	Ex	-
Oils - Vegetable/ Animal	Coconut oil		-	Ex	-
Vegeta Animal	Cod liver oil		-	Ex	-
. Ve	Corn oil		-	Ex	-
- sii	Linseed oil		-	Ex	-
0	Olive oil		-	Ex	-
	Aluminium chloride (dry)	AICI <sub>3</sub> (7446-70-0)	-	Ex	-
	Aluminium sulphate	Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> (10043-01-3)	-	Ex	-
	Alums	(100 10 01 0)	-	Ex	-
	Ammonium bicarbonate	(NH <sub>4</sub> )HCO <sub>3</sub> (1066-33-7)	-	Ex	-
	Ammonium carbonate	(NH <sub>4</sub> ) <sub>2</sub> CO <sub>3</sub> (506-87-6)	-	Ex	-
-	Ammonium chloride	NH <sub>4</sub> Cl (12125-02-9)	-	Ex	-
-	Ammonium monophosphate	NH <sub>4</sub> H <sub>2</sub> PO <sub>4</sub> (7722-76-1)	-	Ex	-
	Ammonium phosphate (dibasic)	(NH <sub>4</sub> ) <sub>2</sub> HPO <sub>4</sub> (7783-28-0)	-	Ex	-
Salts	Ammonium phosphate (tribasic)	(NH <sub>4</sub> ) <sub>3</sub> PO <sub>4</sub> (10361-65-6)	-	Ex	-
S	Ammonium nitrate	NH <sub>4</sub> NO <sub>3</sub> (6484-52-2)	-	Ex	-
-	Ammonium sulfate	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> (7783-20-2) SbCl <sub>3</sub>	-	Ex	-
-	Antimony trichloride	(10025-91-9) BaCO <sub>3</sub>	-	Ex	-
	Barium carbonate	(513-77-9)	-	Ex	-
-	Barium chloride	BaCl <sub>2</sub> (10361-37-2) BaSO <sub>4</sub>	-	Ex	-
	Barium sulfate	BasO <sub>4</sub> (7727-43-7)	-	Ex	-
	Brines	Co/USO /	-	Ex	-
-	Calcium bisulfite	Ca(HSO <sub>3</sub> ) <sub>2</sub> (13780-03-5)	-	Ex	-
	Calcium carbonate	CaCO <sub>3</sub> (471-34-1)	-	Ex	-

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	Calcium chloride		-	Ex	-
	Calcium hypochlorite	Ca(CIO) <sub>2</sub> (7778-54-3)	-	Ex	-
	Calcium sulphate	CaSO <sub>4</sub> (7778-18-9)	-	Ex	-
	Chrome alum	KCr(SO <sub>4</sub> ) <sub>2</sub> (10141-00-1)	-	Ex	-
	Copper acetate	Cu(CH <sub>3</sub> COO) <sub>2</sub> (142-71-2)	-	Ex	-
	Copper chloride	CuCl <sub>2</sub> (7447-39-4)	-	Ex	-
	Copper nitrate	Cu(NO <sub>3</sub> ) <sub>2</sub> (3251-23-8)	-	Ex	-
	Copper sulphate	CuSO <sub>4</sub> (7758-98-7)	-	Ex	-
	Ferric chloride (dry)	FeCl <sub>3</sub> (7705-08-0)	-	Ex	-
	Ferric nitrate	Fe(NO <sub>3</sub> ) <sub>3</sub> (10421-48-4)	-	Ex	-
	Ferric sulfate	Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> (10028-22-5)	-	Ex	-
	Ferrous chloride	FeCl <sub>2</sub> (7758-94-3)	-	Ex	-
pen	Ferrous sulfate	FeSO <sub>4</sub> (7720-78-7)	-	G	-
ontin	Lead acetate	Pb(CH <sub>3</sub> COO) <sub>2</sub> (301-04-2)	-	Ex	-
Salts continued	Magnesium bisulfate	Mg(HSO <sub>4</sub> ) <sub>2</sub> (10028-26-9)	-	Ex	-
S	Magnesium chloride	MgCl <sub>2</sub> (7786-30-3)	-	Ex	-
	Magnesium sulphate (Epsom salt)	MgSO <sub>4</sub> (7487-88-9)	-	Ex	-
	Mercuric chloride	HgCl <sub>2</sub> (7487-94-7)	-	Ex	-
	Mercuric cyanide	Hg(CN) <sub>2</sub> (592-04-1)	-	Ex	-
	Nickel ammonium sulfate	(NH <sub>4</sub> ) <sub>2</sub> Ni(SO <sub>4</sub> ) <sub>2</sub> (7785-20-8)	-	Ex	-
	Nickel chloride	NiCl <sub>2</sub> (7718-54-9)	-	Ex	-
	Nickel nitrate	Ni(NO <sub>3</sub> ) <sub>2</sub> (13138-45-9)	-	Ex	-
	Nickel sulfate	NiSO <sub>4</sub> (7786-81-4)	-	Ex	-
	Potassium aluminium sulphate (potash alum)	KAI(SO <sub>4</sub> ) <sub>2</sub> (10043-67-1)	-	Ex	-
	Potassium bisulfite	KHSO <sub>3</sub> (7773-03-7)	-	Ex	-
	Potassium bromide	KBr (7758-02-3)	-	Ex	-
	Potassium carbonate	K <sub>2</sub> CO <sub>3</sub> (584-08-7)	-	Ex	-

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Moderate	M Suitable for use in environments contaminated by the chemical or in situations where accidental splashing can be removed either by cleaning or in the case of volatile solvents, by evaporation.			
Poor	P Not suitable for any applications involving contact with the chemical itself or fumes evolved from it.			
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	Chemical name (Synonym)	Chemical formula (CAS number)	Concentration	20 °C 68 °F	Other
	Potassium chlorate	KCIO <sub>3</sub> (3811-04-9)	-	Ex	-
	Potassium chloride	KCI (7447-40-7)	-	Ex	-
	Potassium cyanide	KCN (151-50-8)	-	Ex	-
	Potassium dichromate	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> (7778-50-9)	-	Ex	-
	Potassium diphosphate	K <sub>2</sub> HPO <sub>4</sub> (7758-11-4)	-	Ex	-
	Potassium ferricyanide	K <sub>3</sub> [Fe(CN) <sub>6</sub> ] (13746-66-2)	-	Ex	-
	Potassium ferrocyanide	K <sub>4</sub> [Fe(CN) <sub>6</sub> ] (13943-58-3)	-	Ex	-
	Potassium iodide	KI (7681-11-0)	-	Ex	-
	Potassium nitrate	KNO <sub>3</sub> (7757-79-1)	-	Ex	-
	Potassium permanganate	KMnO <sub>4</sub> (7722-64-7)	-	Ex	-
	Potassium sulfate	K <sub>2</sub> SO <sub>4</sub> (7778-80-5)	-	Ex	-
7	Potassium sulfide	K <sub>2</sub> S (1059-82-5)	-	Ex	-
Salts continued	Potassium sulphite	K <sub>2</sub> SO <sub>3</sub> (10117-38-1)	-	Ex	-
s con	Silver nitrate	AgNO <sub>3</sub> (7761-88-8)	-	Ex	-
Salt	Sodium acetate	CH <sub>3</sub> COONa (127-09-3)	-	Ex	-
	Sodium aluminate	NaAlO <sub>2</sub> (1302-42-7)	-	Ex	-
	Sodium bicarbonate	NaHCO <sub>3</sub> (144-55-8)	-	Ex	-
	Sodium bisulfate	NaHSO <sub>4</sub> (7681-38-1)	-	Ex	-
	Sodium bisulfite	NaHSO <sub>3</sub> (7631-90-5)	-	Ex	-
	Sodium borate (borax)	Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub> (1303-96-4)	-	Ex	-
	Sodium bromide	NaBr (7647-15-6)	-	Ex	-
	Sodium carbonate (soda ash)	Na <sub>2</sub> CO <sub>3</sub> (497-19-8)	-	Ex	-
	Sodium chloride	NaCl (7647-14-5)	-	Ex	-
	Sodium chromate	Na <sub>2</sub> CrO <sub>4</sub> (7775-11-3)	-	Ex	-
	Sodium cyanide	NaCN (143-33-9)	-	Ex	-
	Sodium fluoride	NaF (7681-49-4)	-	Ex	-

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Good	G	Suitable for applications involving immersion for short periods, splashing and contact with fumes.	
Moderate	М	Suitable for use in environments contaminated by the chemical or in situations where accidental splashing can be removed either by cleaning or in the case of volatile solvents, by evaporation.	
Poor	Р	Not suitable for any applications involving contact with the chemical itself or fumes evolved from it.	
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FN 10182



	Chemical name (Synonym)	Chemical formula (CAS number)	Concentration	20 °C 68 °F	Other
	Sodium hypochlorite (bleach)	NaClO (7681-52-9)	15%	М	-
	Sodium metaphosphate	(NaPO <sub>3</sub> ) <sub>6</sub>	-	Ex	-
	Sodium metasilicate (sodium silicate)	Na <sub>2</sub> SiO <sub>3</sub> (6834-92-0)	-	Ex	-
	Sodium nitrate	NaNO <sub>3</sub> (7631-99-4)	-	Ex	-
per	Sodium phosphate (dibasic)  Na <sub>2</sub> HPO <sub>4</sub> (7558-79-4)		-	Ex	-
continued	Sodium phosphate (tribasic)  Na <sub>3</sub> PO <sub>4</sub> (7601-54-9)		-	Ex	-
Salts co	Sodium sulfate	Na <sub>2</sub> SO <sub>4</sub> (7757-82-6)	-	Ex	-
Sa	Sodium sulfide	Na <sub>2</sub> S (1313-82-2)	-	Ex	-
	Stannous chloride (tin chloride)	SnCl <sub>2</sub> (7772-99-8)	-	Ex	-
	Zinc chloride ZnCl <sub>2</sub>		-	Ex	-
	Zinc hydrosulfite	ZnS <sub>2</sub> O <sub>4</sub> (7779-86-4)	-	Ex	-
	Zinc sulfate	ZnSO <sub>4</sub> (7733-02-0)	-	Ex	-

Excellent	Ех	Suitable for all reasonable applications including immersion.	
Good	G	Suitable for applications involving immersion for short periods, splashing and contact with fumes.	
Moderate	М	Suitable for use in environments contaminated by the chemical or in situations where accidental splashing can be removed either by cleaning or in the case of volatile solvents, by evaporation.	
Poor	P	Not suitable for any applications involving contact with the chemical itself or fumes evolved from it.	
*		Product must be post cured to deliver quoted chemical resistance	

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